

Techniques in Molecular Genetics

2010 edition

H.E. Schellhorn

Day 1

- Introduction
 - Why are we here?
 - Overview
 - Teaching Coordinator, Teaching Assistants
 - Changes for 2010
 - Techniques
 - Use of a Pipetman
 - Streak a culture, Make some media
 - DNA Management Software

Course Rationale

This course is primarily aimed at students who are starting to work in molecular biology research mainly Biology and Molecular Biology student who have completed third year and are working in the Biology department during the summer. The formal part of the course, consisting of two weeks of laboratory/lecture, runs the first two weeks of May. The objective is to provide participants with formal instruction in the scientific process including laboratory techniques that they need to accomplish their research objectives. By combining theory with practice, much duplication in instruction among labs will be eliminated.

Instructors



**Dr. H.E.
Schellhorn**



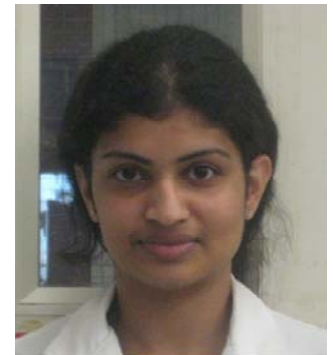
**Ms. Alison
Cowie**



Jas Wasniewski



Sarah Chaing



**Sharmila
Sathiasothy**

Significant Changes for 2010

- Safety lessons converted to assignments.
- Lab Notebook 20% rather than 30%.
- New Assignment, to be given on May 14th, work 10%.

Grading Scheme

Quiz I	20%	(May 7th, days 1-4)
Quiz II	30%	(May 14th, days 1-9)
Overnight and Rotation Assignments	10%	
Course performance	10%	
Lab notebook keeping	20%	
Assignment (last day)	10%	

Use and understand the principle of the following laboratory equipment/tools...

- Laboratory notebook.
- Centrifuge.
- Spectrophotometer.
- Image analysis.
- Scanner.
- Scintillation counter.
- Autoclave.
- pH meter.
- PCR cycler.
- Transilluminator.
- Balance.
- Analytical balance.
- Sonicator.
- Gel dryer.
- Computer.
- Web tools.
- Spreadsheets

Potential Overnight assignments..

- Use Refworks to write a short essay...
- Write an SOP/AUP
- Read “instructions to authors” and answer a short quiz
- Prepare a table comparing protein methods
- Prepare a table comparing graph types
- Prepare a publication quality graph
- Prepare an order sheet for purchase of chemicals
- Write and submit Primer, DNA sequencing order

Wiki/Website

- Part will be public..some parts will require a login in.
- Include product manuals (PDFs), assay manuals (PDFs) reference tables, calculators and sample spreadsheets.
- Will also include web resource for each technique.

Practical

- Why do experiments fail?
- How to plan experiments
- Where to store samples.

4XX3: Lab Rules and Organization

NO FOOD

Lab coat: General safety, and biosafety level 2 tissue culture work

Safety equipment: Fire, Eye wash, shower

Safety goggles: for acid/base handling, fume hood for HCl, SDS, BME

Gloves: for handling of acrylamide, acid, basis, ethidium bromide
biosafety level 2 tissue culture work

Clean balances!!!

Waste: biological waste vs non-biological waste

Reasoning

- Scientific method
- The falsifiable hypothesis-Popper
- What makes a good (powerful) hypothesis?
- Induction/deduction Reasoning (specific to the general)
- Predictive models
- Theory/Proven fact
- Cause vs correlation
- Reductionist/holistic
- Science/Magic
- Orthogonality
- Conjecture vs plausible explanation

Ethics

- Plagiarism—degrees of plagiarism, recent examinations of the problem
- Fraud
- Accreditation
- Misunderstanding

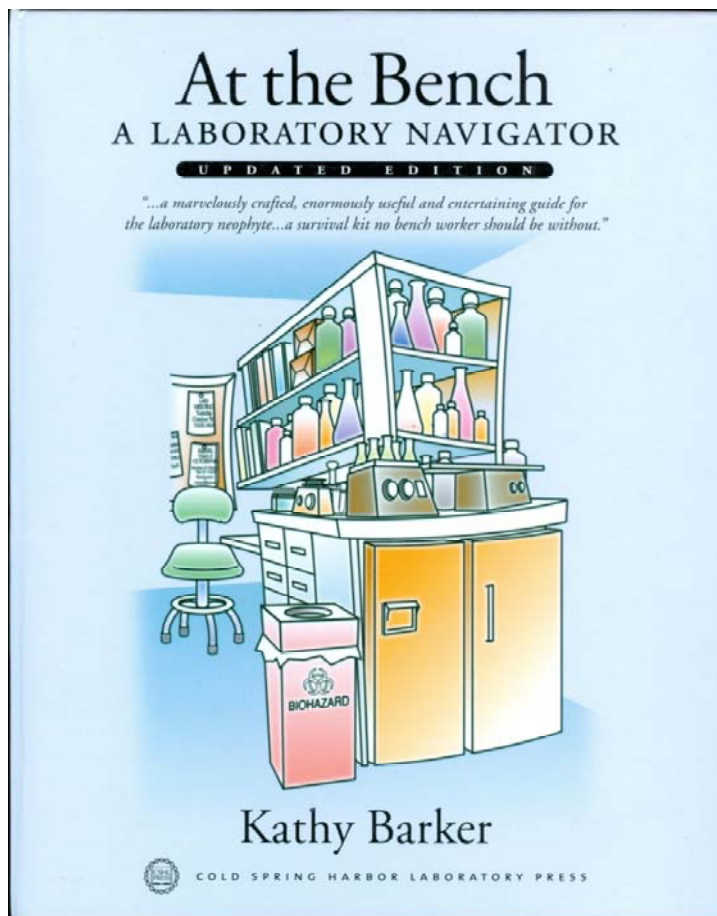
The 10 most common mistakes made in laboratory research. (HES)

1. Failing to promptly write up experiments and write out protocol before hand.
2. Failing to include the proper controls.
3. Not preparing enough material.
4. Failing to store properly store material promptly.
5. Allowing a distraction to screw up the addition of a key reagent.
6. Improper (usually insufficient) mixing or agitation. esp. frozen reagents.
7. Not discussing results with you supervisor/colleague before proceeding to the next step.
8. Not checking the accuracy of pipettor/pH meter (or other instrument....) before assay.
9. Calculation error made in making up reagent (factors of ten/failure to take into account water of hydration in calculation).
10. Calculation error made in determination of results.
11. Not labelling tubes/dishes etc.

Documentation

- keep a copy of all protocols
- must be able to document **primary** literature references
- must include details of all experiments
- must have original protocol and your own protocol that is sufficient to allow another person to reproduce the experiment

Recommended book



There are very few books that described the laboratory environment for new researchers. “At the Bench” is probably one of the best and I recommend that you either buy or, perhaps preferably, ask you supervisor to but it for the lab.

1

General Lab Organization and Procedures

WELCOME TO ONE of the most exciting and enjoyable workplaces ever evolved, the biomedical research laboratory. There is an amazing concept in operation here: You get paid or get credit for doing experiments, surely an almost scandalously delightful way to make a living. The work is worthwhile. The dress code, if any, is casual. The work hours are often self-determined and based on the needs of the experiment. The lab or department is filled with bright and interesting people with whom you can discuss the salt concentration needed for a kinase assay or the implications of the latest congressional bill. It can come to have all the psychological comforts of home.

Like any complex social organizations, research laboratories have their own customs and rules. The difficulty is that the rules have been unspoken. You are expected to decipher the many obtuse clues and become a law-abiding member of a society in which individualism is highly prized. Although no one is expected to show you how to work the equipment, you will be expected to work it. In a profession in which communication of data is the goal and the reward of the research, not all people can communicate with you clearly and satisfactorily. Don't worry, you will manage! In a short time, the pleasure of working together with colleagues on interesting and similar projects will supplant any initial feelings of unease. But to get your work done well, you must first navigate among sometimes vague and mixed signals and learn how your laboratory beats and hums.

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Laboratory tasks, lab jobs, assigned jobs	9
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Presentations

- Basic area - why is it of general interest?
- Background relevant observations
- Hypothesis being tested - make a clear statement
- Describe basic experimental approach including methodology and alternative methodology
- Present results - describe the results and compare, when appropriate, to your own work and to literature results

Dissemination of Results

before computers

- rough results presented first and results for publication drafted later

today

- publication can be made directly from raw data

For the student:

- standardized instruction
- familiarity with departmental equipment
- appreciation for expectation of performance
- coupling of theoretical with practical training
- can concentrate on practical material without the distraction of other courses

For the professor

- reduction in need for instruction
- increased productivity of student during the summer..

Supervisor Expectations...

- Read the literature
- Ask questions (be inquisitive..)
- Keep regular hours (at least until you publish a paper..)
- Give lab talks (with enthusiasm..)
- Produce final tables/figures
- Organize data—cross references to computer files
- Properly store samples (archival if necessary)
- Properly dispose of samples.
- ***Be Intellectually Engaged!!!!***

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Student Expectations...

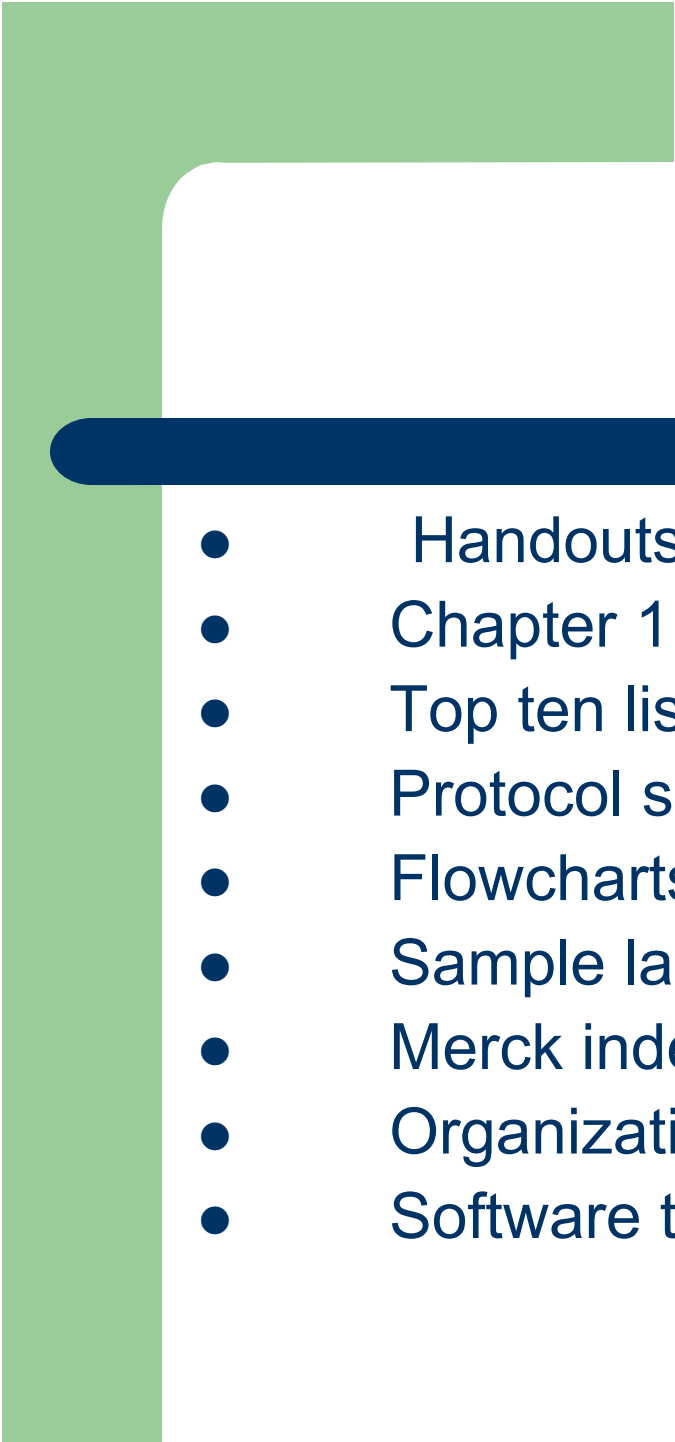

- Reference Letter
- Valuable Experience
- A Paycheck..

Materials

- My lab page Forsbergs lab
- At the bench
- Gilson Guide to Pipetting
- Protocol sheet
- Handouts

Thesis

- Use “Table of Contents” function in Word
- Use Reference Manager or equivalent (Refworks)
- Protocols (Standard Operating Procedure (SOP))
- Appendices (raw data)

- 
- 
- Handouts
 - Chapter 1
 - Top ten list
 - Protocol sample
 - Flowcharts
 - Sample lab book
 - Merck index
 - Organization of data in spreadsheet
 - Software tools

Laboratory notebook

1. Title: e.g. Lab #1 - Microscopy and Examination of Living and Stained Cultures
2. Introduction: Briefly state why this experiment was conducted in your own words – do not copy the lab manual.
3. Objective: Briefly state what you are attempting to determine
4. Materials and Methods: If same as the manual then refer to the lab manual. If different from the manual state the differences. For your own understanding use flow charts to illustrate procedures.
5. Results: If possible, use table(s) and/or figure(s) to present raw data. **Provide brief descriptions of what the data mean.**
6. Discussion: Briefly discuss what you can conclude from your results. Sometimes experiments fail either because of an unanticipated variable or because of experimental error. If your results deviate from expectations, identify possible sources of error, provide alternative hypotheses, and suggest improvements for future experiments.
7. Questions: Briefly answer the questions given at the end of each lab.
8. Fill in the table of contents to permit easy orientation.
9. Sign and date each page. Signature of lab partners or advisors must be included. However, this is an essential procedure in government and industry labs.
10. Answer the Lab Rotation questions on the pages indicated.

Linus Pauling (Nobel Laureate) maintained labnotes....

5 March 1954 Dichroism of collagen for Amide II vibration 41

Prof. Badger has told me today that the ~~Amide II~~ ^{NH} vibration in glycylglycine lies about 10° from N-H, toward N...O, and that the Amide II vibration is very nearly perpendicular to it, and in the plane of the group.

I find that the C-C-C axis is perpendicular to a line 135° from C-N, and hence 12° out from N-H (assuming it to be 123° from C-N). Hence the dichroism can be calculated from values of $z_i - z_j$ for C's, using 3.83 \AA for C₁₁-C_i.

	Δx	$(\frac{\Delta x}{3.83})^2 = \mu_z^2$	$\mu_x^2 = \mu_y^2$	μ_z^2	$\mu_x^2 = \mu_y^2$	
C-C ₁	2.08	0.296	0.352	0.296	0.352	This calculation is made because 2/3 of N ₂ have C's attached, & give no Amide II frequency. (This may not be correct - check it with Badger 1954)
C-C ₂	2.30	.926	.297	0.141	0.096	
C-C ₃	2.48	.420	.290	0.410	0.290	
		1.172	0.947	0.558	0.738	

Dichroism = $\frac{0.952}{0.738} = 1.16$, and positive.

Badger says he observes ~ 1.22 positive, for collagen.

6 March 1954
A. Let us calculate, for a single-chain all-trans structure, the minimum amount of negative course (minimum value of $z_{i+1} - z_i$, dC, for one residue.

The second residue, N₂C₂, is 70% occupied by pro + hyper.

The value of $\cos \theta_1 + \cos \theta_2 + \cos \theta_3 = 0.75$. ($= 2.57/3.83$).

If there is no retrograde character, the maximum dichroism (positive) is obtained by putting $\cos \theta_1 = 0.75$ (pro), $\pm \cos \theta_2 = \cos \theta_3 = 0$. This gives $0.56 / (0.44 + 1.3) = 0.56 / 0.87 = 0.64$. i.e., negative dichroism.

\therefore An acceptable structure must have retrograde character.

The minimum retrograde character is obtained by putting it all in $\cos \theta_1$, and all return in $\cos \theta_3$. This gives, with $\cos \theta_1 = x$

$$\frac{x^2 + (x - 0.75)^2}{2(1 - x^2 + 0.70 + (1 - (x - 0.75)^2))} = 1.22 \quad \text{or } x = 0.915$$

Hence there must be retrograde character by at least $(0.915 - 0.75) 2.90 \text{ \AA} = 0.49 \text{ \AA}$

Aside from I c (10,10,2), the only three-residue two-H-bonded single-chain all-trans structures with pitch $\sim 2.9 \text{ \AA}$ /unit are III A (16,16,2), III B (16,10,2), and III C (14,14,2). None of these three is retrograde; \therefore they are eliminated.

<http://osulibrary.oregonstate.edu/specialcollections/rnb/>



Linus Pauling Research Notebooks

As with many scientists, Linus Pauling utilized bound notebooks to keep track of the details of his research as it unfolded. A testament to the remarkable length and diversity of Dr. Pauling's career, the Pauling Papers holdings include forty-six research notebooks spanning the years of 1922 to 1994 and covering any number of the scientific fields in which Dr. Pauling involved himself. In this regard, the notebooks contain many of Pauling's laboratory calculations and experimental data, as well as scientific conclusions, ideas for further research and numerous autobiographical musings.

Research Notebook 01

1922

Research Notebook 02

1922-1923, 1932, 1934, 1936, 1973, 1985

Research Notebook 03

1923-1925

Research Notebook 04

1923-1924, 1928-1930

Research Notebook 05

1924, 1929, 1933, 1935

Research Notebook 06

1929-1930, 1934

Research Notebook 07

1930, 1932-1934, 1936

Research Notebook 08

1930-1931, 1933, 1935-1936

Research Notebook 09

1932, 1934-1938, 1940-1941

Research Notebook 10

1933-1934, 1936, 1938

Research Notebook 13

1935-1936, 1938-1939

Research Notebook 14

1936-1939, 1949, 1952

Research Notebook 15

1935, 1937, 1968

Research Notebook 16

1935-1956

Research Notebook 17

1939-1941, 1971, 1988

Research Notebook 18

1936, 1938, 1955, 1967-1969, 1971-1975, 1978

Research Notebook 19

1941-1942, 1944, 1949, 1955-1956, 1959-1960, 1965, 1967, 1976-1978, 1980-1981

Research Notebook 20

1950, 1955-1960

Research Notebook 21

1958

Research Notebook 24

1953, 1956, 1962, 1963, 1967, 1968, 1969, 1970, 1973

Research Notebook 25

1958, 1964-1966

Research Notebook 26

1955, 1964-1969, 1974-1976, 1980-1982, 1987, 1990-1991

Research Notebook 27

1952-1954, 1960-1961, 1964, 1971-1972

Research Notebook 28

1951, 1953-1957, 1972-1975

Research Notebook 29

1957, 1986-1988

Research Notebook 30

1954-1958

Research Notebook 31

1966-1971, 1974-1976, 1980

Research Notebook 32

1965-1968, 1974

Research Notebook 35b

1938-1939, 1946, 1955, 1968, 1986-1988

Research Notebook 36

1980-1981, 1986-1987

Research Notebook 37

1971, 1983

Research Notebook 38

1980-1981, 1983, 1985, 1989

Research Notebook 39

1980-1981

Research Notebook 40

1988-1989

Research Notebook 41

1989-1990

Research Notebook 42

1990

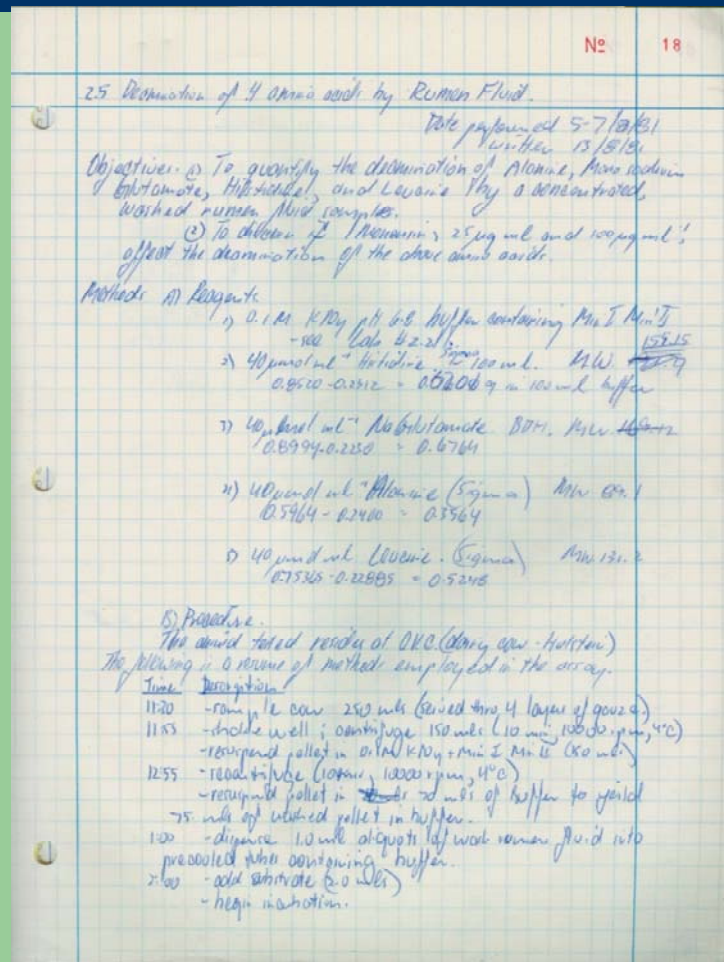
Research Notebook 43

1990-1991

Research Notebook 44

1991-1994

My first lab book....



It is more important that notebooks be complete than neat (although that latter is nice too...).

How spreadsheet can be adapted to notebooks...

Date May 6, 2008

Purpose: To examine growth phase dependent reporter gene expression in wild type (GC4466) and a mutant strain (H5143) carrying an operator fusion to the *trpB* promoter.

Materials: Strains E. coli GC4466 (WT)
E. coli H5143 (trpB-mc2)
LB broth
0.4mg/ml ONPG (β -galactosidase substrate)
Biorad Coomassie Blue in Phosphoric acid (protein dye)

Preparation of cultures
-strains were grown overnight in LB from single colonies in 5.0 ml LB broth
-subcultured (100 μ l in 50ml) the next morning and grown for 3-4 h in an ODB000v4.3 (exponential phase)
-5.0 ml overnight culture collected (stationary phase)
-growth was stopped by the addition of 100 μ g/ml chloramphenicol (final conc.)

Preparation of cell extracts
-cells were washed 2 x with 0.25M KPI buffer pH 7.0 (50mg, 4C, 15 min) in a Biorad 800000 telescope centrifuge
-cultures were resuspended in 400 μ l KPI buffer pH 7.0 in 15 ml polypropylene tubes
-cell suspensions then disrupted by sonication 20 min (20 x on, 20 x off, full setting) in Heat Systems sonicator equipped with a cuphorn
-100 μ l of culture were then transferred to 1.5 microtiter tubes and centrifuged (1400g, 4C, 15 min) in 11.0 ml cold rotor
-supernatant (cell free extracts CFE) were then transferred to new microtiter tubes for enzyme assay and PAGE

Determination of protein
-the Bradford (1976) assay was used to measure protein employed the Biorad Micro assay.
-Biorad Micro assay
-place 1.0 μ l sample in 1.0 ml water
-add 200 μ l Biorad Coomassie Blue in Phosphoric acid using a repeating pipettor
-read absorbance at 565 nm

D standard curve was prepared using a 1.0 mg/ml bovine serum albumin standard
-Biorad Coomassie Blue dye reagent (was a lab read as this reagent contains concentrated phosphoric acid)

Sample data

Standard Curve	BSA Prot. μ g	Rep 1	Rep 2	Rep 3	Average	Average Absorbance
1.0	0.0	0.294	0.282	0.296	0.291	0.000
4.0	0.526	0.513	0.539	0.528	0.512	
8.0	0.505	0.490	0.504	0.490	0.494	
12.0	0.721	0.722	0.72	0.721	0.722	
16.0	0.834	0.837	0.83	0.834	0.839	

1.0/4

Regression Output:
Constant: 0.048
Std Err of Y Est: 0.019172
R Squared: 0.990251
No. of Observations: 5
Degrees of Freedom: 3
X Coefficient(1): 0.02668
Std Err of Coef.: 0.001516

Strain	Phase	Sample size μ l	Rep 1	Rep 2	Rep 3	Average	Protein μ g/ml
GC4466	Exponential	1.0	0.8000	0.7200	0.765	0.788	7.3
	Stationary	1.0	0.8020	0.8000	0.786	0.796	7.8
H5143	Exponential	1.0	0.7110	0.7430	0.726	0.726	6.1
	Stationary	1.0	0.7380	0.720	0.718	0.726	7.2

Standard Curve

BSA Prot. μ g	Rep 1	Rep 2	Rep 3	Average	Average Absorbance
0.0					
4.0					
8.0					
12.0					
16.0					

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Strain	Phase	Sample size μ l	Rep 1	Rep 2	Rep 3	Average	Protein μ g/ml
GC4466	Exponential	1.0					
GC4466	Stationary	1.0					
H5143	Exponential	1.0					
H5143	Stationary	1.0					

Determination of β -galactosidase
-samples were assayed for β -galactosidase using the method of Miller (1972)
-place 1.0 ml 2 buffer pH 7.0 into labeled disposable microtiter tubes in triplicate
-add 1.0 μ l of supplied cell extract to tubes and mix gently
-add 200 μ l of 0.4mg/ml ONPG substrate, begin timing
-stop reaction by the addition of 0.5 ml of 1.0 M Na₂CO₃
-read optical density at 420nm and record results
-assume the molar extinction coefficient of ONPG is 4.5x10³ l.mol⁻¹.cm⁻¹

Strain	Phase	Sample size μ l	Time of assay	Rep 1	Rep 2	Rep 3	Average	nmol	Protein μ g/ml	Specific Activity nmol min ⁻¹ mg ⁻¹
GC4466	Exponential	1.0	2:50							
GC4466	Stationary	1.0	1:30							
H5143	Exponential	1.0	1:40							
H5143	Stationary	1.0	1:45							

Blank

Conclusions
Start \rightarrow 12:35

Handwritten notes:
tube ① GC4466 (E)
② GC4466 (S)
③ H5143 (E)
④ H5143 (S)
1.0/4
1.0 μ g/ml

Reference Texts

- SOPs and AUPs
- Merck Index
- Biochemical Data Book
- Maniatis et al